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These lecture presentations were designed for my high school Chemistry I Honors class. Students of high school and college general chemistry may find them useful as a supplement to their own class notes or as a review. Teachers, please feel free to use and modify them for your own classes.

Mrs. J's Chemistry Page - Lecture Notes

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Chemistry End of Chapter Exercises. Write a brief description or definition of each of the following: (a) nucleon (b) α particle (c) β particle (d) positron (e) γ ray (f) nuclide (g) mass number (h) atomic number. Which of the various particles (α particles, β particles, and so on) that may be produced in a nuclear reaction are actually ...

21.2 Nuclear Equations – Chemistry

Nuclear fusion: It is a type of reaction where two or more elements fuse together to form a larger element, releasing a large amount of energy in the process. To learn more about nuclear chemistry topics like nuclear fusion and more, register with BYJU'S and download our app.

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CHAPTER 3 - Nuclear Mass and Stability. Pages 41-57. Select CHAPTER 4 - Unstable

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