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Reaction Periodic Trends:

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Chapter 7 Notes - Atomic
Structure and Periodicity .
7.1 Electromagnetic
Radiation . A. Types of EM
Radiation (wavelengths in
meters) 10^{-12} . 10^{-10} . 10^{-8} .
4 to 7×10^{-7} . 10^{-4} . 10^{-2} . 1
10. 2. 10. 4. Wavelength

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increases . Frequency
decreases . Energy decreases
. Speed is constant = 2.9979×10^8 m/sec . B. Properties
of EM Waves 1.

Chapter 7 Notes - Atomic Structure and Periodicity

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nterconvert wavelength and
frequency 2:18; Calculate
the energy of a photon from

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its wavelength 3:19; Find the energy change and wavelength of the photon absorbed or emitted when an H atom's electron changes energy level. ...

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Match. Gravity. Created by.
APMother. Zumdahl Sixth
edition. Terms in this set
(36) electromagnetic
radiation. radiant energy

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that exhibits wavelike
behaviour and travels
through space at the speed
of light in a vacuum.
wavelength.

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copy of the periodic table
on Monday, November 16th.
Please have questions ready
for a review on Monday.

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Review Questions Multiple-choice exercise. Use these values to assist you in answering some of the questions in this problems set: $h = 6.626 \times 10^{-34} \text{ J}\cdot\text{sec}$
 $c = 2.9979 \times 10^8 \text{ m/sec}$.
Show all questions <= => An

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element having the
configuration [Xe]6s 1
belongs to the Group: ?

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Aufbau Principle a) depicts the atom as a small, positively charged nucleus surrounded by electrons that travel... 2) A ray of Ultraviolet light has a wavelength of 50nm.

Calculate the Frequency of

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this ray of light a)
12,000,000 Hz b)... 3) The
only 2 possible Ms numbers
in ...

Chapter 7 Quiz - AP

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A.P. Chemistry Practice Test

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- Ch. 7, Atomic Structure and Periodicity MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question. 1) Ham radio operators often broadcast on the 6-meter

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band. The frequency of this electromagnetic radiation is

A) 2.44×10^{16} B) 1.04×10^{-13}
C) 9.62×10^{12} D) 3.69 E) 4.10×10^{-17}

**A.P. Chemistry Practice Test
- Ch. 7, Atomic Structure**

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and . . .

Stoichiometry and molecular composition. Balancing chemical equations. : Stoichiometry and molecular composition. Stoichiometry. : Stoichiometry and molecular composition.

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Limiting reagent
stoichiometry. :
Stoichiometry and molecular
composition. Molecular
composition. : Stoichiometry
and molecular composition.

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STUDY. PLAY. Electromagnetic radiation. radiant energy that exhibits wavelike behavior and travels through space at the speed of light in a vacuum. Wavelength. the

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distance between two consecutive peaks or troughs in a wave. Frequency.

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Chapter 7 (Atomic Structure and Periodicity) - Part 4

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55:02 The First Exoplanet
From Outside Our Galaxy -
The Secret of Our Solar
System How the Universe
Works 214...

Chapter 7 (Atomic Structure

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and Periodicity) – Part 3

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terms. Chapter 6 Vocab

Chemistry. 43 terms. Chapter

4 VOCAB. 26 terms. chem ch

4. OTHER SETS BY THIS

CREATOR. 11 terms. Irregular

Past Participle. 80 terms.

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Capitulo 3 Vocabulario. 11
terms. Verbs for
requests/demands/wishes. 67
terms. Capitulo 2
Vocabulario. THIS SET IS
OFTEN IN FOLDERS WITH...

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AP Chemistry; Ch 1 and 2:
Scientific Notation and Unit
Analysis. Matter Handout.
Time Problems Dimensional
Analysis Dimensional
Analysis Problems. ...
Chapter 15 Acid Base

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Answers. Ch 15 Handout Keys
. Ch 16 Titration
Curves/Calculations Answers
Ch 16 Acid/Base Practice

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Structure and Periodicity
7.1 Electromagnetic
Radiation A. Types of EM
Radiation (wavelengths in
meters) 10^{-12} 10^{-10} 10^{-8} 4
to 7×10^{-7} 10^{-4} 10^{-2} 1 10^2

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104 Wavelength increases
Frequency decreases Energy
decreases Speed is constant
 $= 2.9979 \times 10^8 \text{ m/sec}$ B.
Properties of EM Waves 1.
Wavelength (l) a.

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Page 42/49

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| CourseNotes

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Section 6.4 (Do not complete
the following: 6.4 Activity:
The Colors of Chemical
Bonds) Chapter 6.5 will be
completed in class with Mrs.
Gaines; AP Chemistry 1

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Textbook: Section 6.6

(Compete in Class: Activity
6.7: Modeling the AX_mEn
Molecules) AP Chemistry 1
Textbook, section 6.7

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Electron Configuration
Atomic Orbitals Definition-
the distribution of
electrons amongst the atomic
orbitals Pauli Exclusion
Principle- No two electrons
in the same atom can have
the same set of four quantum

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numbers Dimagnetism-
Repelled by a magnet, no net
unpaired spins

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Page 46/49

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Questions Multiple-choice
exercise. Use these values
to assist you in answering
some of the questions in
this problems set: $h = 6.626$
 $\times 10^{-34}$ J·sec $c = 2.9979 \times$
 10^8 m/sec. Show all
questions $\leftarrow \rightarrow$ An element

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having the configuration
[Xe]6s 1 belongs to the
Group: ? halogens ? noble
gases ? alkali

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